

Gas Leak on Converter 2nd Pass Outlet

$P/P_A = 1.356$ less than 1.893 $[(k + 1) / 2]^{k / (k - 1)}$ therefore the flow is non-choked (i.e. subsonic), AND the following equation applies

$$Q = CAP \sqrt{(2g_c / ZRT)(K/K-1)[(P_A/P)^{2/K} - (P_A/P)^{(K+1)/K}]}$$

Q = mass gas flow (lbs/s)

C = discharge coefficient

Equivalent Diameter of hole (in)

A = area of hole (ft²)

g_c = gravitational constant (ft/s)

R = gas constant (ft-lb/lb mol - °R)

T = temperature (°R)

Molecular weight of SO₃

0.65

0.77

0.00323

32.17

1543.3

1410

80

K = C_p/C_v of the gas

P = source pressure absolute (lb/ft²)

P_A = ambient pressure absolute (lb/ft²)

M = molecular weight of gas

Z = compressibility factor

Release duration (seconds)

SO₃ concentration in gas (wt%)

SO₂ concentration in gas (wt%)

Molecular weight of SO₂

1.4

2871

2117

34

1.077063

35,496

22.4

2.0

64

psia

psig

in WC

Intermediate Calculations:

0.000934

3.5

0.647034

0.593079

6.031606

Mass Calculations:

Q =

0.0801 lbs/s

Total mass:

2843 lbs

Total SO₃ mass:

318 lbs


Total SO₂ mass:

28 lbs

Note: SO₃ immediately reacts with H₂O to form acid mist; RQ for acid mist = 1,000 lbs

Reference: "Perry's Chemical Engineering Handbook, 6th Edition, McGraw-Hill 1984"

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| EXHIBIT # | 1 |
| DEPONENT | Monhollen |
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